

NEW SIMPLE AND ONE-POT SYNTHETIC ROUTES TO POLYFUNCTIONALLY SUBSTITUTED IMIDAZO- [1,2-*a*]PYRIDINES, PYRIDO[1,2-*a*]PYRIMIDINES, PYRIDO- [1,2-*a*]-1,3-DIAZEPINES, AND IMIDAZO[1,2-*a*]PYRIMIDINES

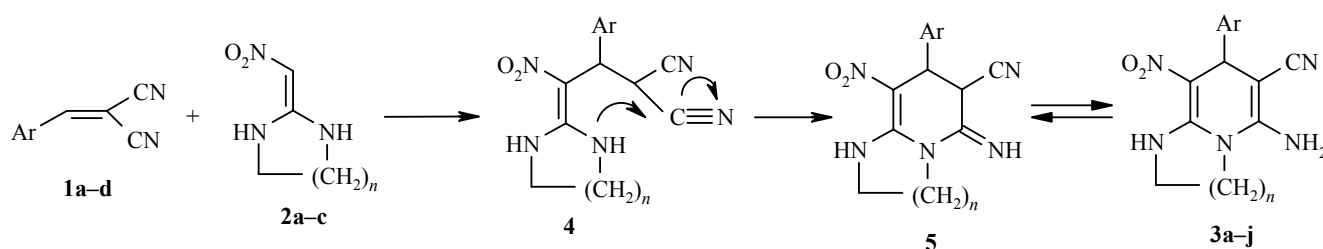
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*Reactions of arylidenemalononitriles with 2-nitromethylene-substituted imidazolidine, hexahydropyrimidine, and hexahydro-1,3-diazepine afforded the title derivatives. Reaction of 2-nitromethylenehexahydropyrimidine with benzylamine or *n*-butylamine and formalin in a molar ratio of 1:1:2 gave the hexahydro-1H-imidazo[1,2-*c*]pyrimidine derivatives. Treatment of 2-nitromethyleneimidazolidine and hexahydropyrimidine with secondary aliphatic amines and formalin in a molar ratio of 1:1:1 afforded the corresponding methylene bis-compounds.*

Keywords: arylidenemalononitriles, nitroenamines, Mannich reaction

In light of the reported biological activity of some imidazo[1,2-*a*]pyridines [1, 2], and pyrido[1,2-*a*]pyrimidine [3] and in continuation of our program aiming to synthesize polyfunctionally substituted heterocycles of potential biological activity [4-8], it was interesting to study the reactions of arylidenemalononitriles **1a-d** with some nitroenamines **2a-c** as a new synthetic route to polyfunctionally substituted imidazo-[1,2-*a*]pyridines, pyrido[1,2-*a*]pyrimidines, and pyrido[1,2-*a*]-1,3-diazepines.

Scheme 1



2 a *n* = 1, **b** *n* = 2, **c** *n* = 3; **3 a** Ar = Ph, **b, e, h** Ar = C₆H₄Cl-*p*, **c, f, i** Ar = C₆H₄Br-*p*,
d, g, j Ar = C₆H₄NO₂-*p*; **a-d** *n* = 1, **e-g** *n* = 2, **h-j** *n* = 3

Thus, the reaction with 2-(nitromethylene)imidazolidine (**2a**) [9], hexahydro-2-(nitromethylene)pyrimidine (**2b**) [10], and/or hexahydro-2-(nitromethylene)-1H-1,3-diazepine (**2c**) [11] in a molar ratio of 1:1:1 in refluxing ethanolic piperidine afforded imidazo[1,2-*a*]pyridines **3a-d**, hexahydro-

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pyrido[1,2-*a*]pyrimidines **3e-g**, and hexahydro-1H-pyrido[1,2-*a*]-1,3-diazepines **3h-j**, respectively. The structures of **3** were established for the reaction products, based on their analytical and spectral data (see Experimental).

Formation of **3** can be explained *via* the initial Michael addition of **2a-c** to the ylidenic bond in **1**, leading to the formation of an acyclic intermediate **4** which cyclized into the intermediate **5** *via* nucleophilic attack of an NH group on a cyano carbon, followed by tautomerization to the final product **3**.

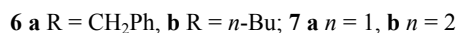
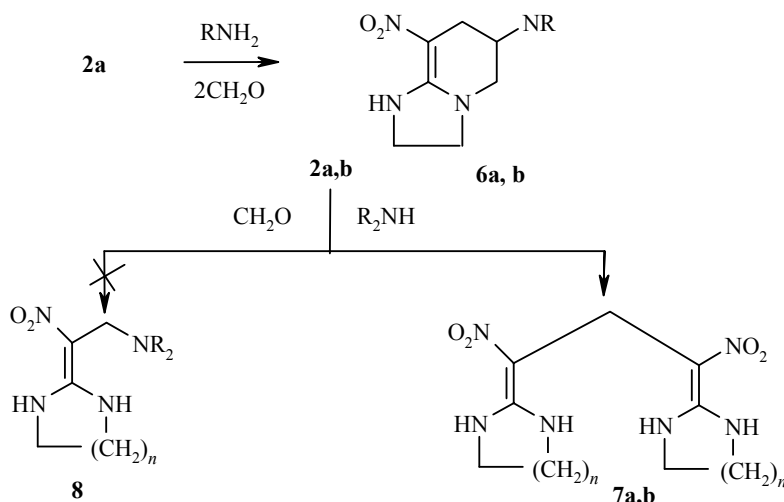
Compounds **3a-j** were also synthesized by another route involving one-pot condensation of the appropriate aromatic aldehyde, ArCHO, malononitrile, and compounds **2a-c** in a molar ratio of 1:1:1 in ethanolic piperidine.



In this case, formation of **3** is interpreted in terms of the initial condensation of aldehyde with malononitrile affording the activated cyano olefin **1**, followed by addition of **2** to **1** as above.

It has been reported that acyclic N-monosubstituted nitroenamines react with primary amines and 2 mol of formaldehyde to yield tetrahydropyrimidines [12]. It appeared of interest to study the behavior of **2a,b** towards a double Mannich reaction as a route to condensed pyrimidine systems.

Thus, when **2a** was subjected to reaction with benzylamine or *n*-butylamine and formalin in a molar ratio of 1:1:2 it afforded hexahydro-1H-imidazo[1,2-*a*]-pyrimidine derivatives **6a,b**. The products **6a,b** were characterized by elemental analyses and spectral data (cf. Experimental).



Attempts to synthesize pyrimido[1,2-*c*]pyrimidine from **2b** ($n = 2$) under the same reaction conditions have been unsuccessful. On the other hand, treatment of **2a,b** with formalin and secondary aliphatic amines in a molar ratio of 1:1:1 under Mannich reaction conditions afforded methylene bis-compounds **7a,b** instead of the expected products **8**. The formation of **7** could be rationalized through a mechanism which involves the formation of **8** as intermediate followed by attack on nitroenamines **2a,b**.

EXPERIMENTAL

All melting points are uncorrected. Elemental analyses were performed in the Microanalytical Unit, Mansoura and Cairo University. IR spectra were recorded in pellets of KBr on a Pye-Unicam spectrophotometer. ^1H NMR spectra are obtained on Bruker 250 at 200 MHz using TMS as an internal standard. Mass spectra were obtained using GC-MS QP 1000 EX Shimadzu, or HP model: MS-5988 mass spectrometer at 70 eV.

5-Amino-7-aryl-6-cyano-8-nitroimidazo[1,2-*a*]pyridines (3a-d), 6-Amino-8-aryl-7-cyano-9-nitro-1,2,3,4,5,8-hexahydropyrido[1,2-*a*]pyrimidines (3e-g) and 7-Amino-9-aryl-8-cyano-10-nitro-2,3,4,5,6,9-hexahydro-1H-pyrido[1,2-*a*]-1,3-diazepines (3h-j) (General Procedure). A. A mixture of **2a-c** (0.003 mol) and the appropriate arylidenemalononitriles **1a-d** (0.003 mol) in absolute ethanol (25 ml) and piperidine (0.1 ml) was refluxed for 4 h. The solid products deposited or obtained after concentration and cooling were filtered off and crystallized from the proper solvent to give compounds **3a-j**.

B. A mixture of **2a-c** (0.003 mol), the appropriate aldehyde (0.003 mol), and malononitrile (0.003 mol) in absolute ethanol (25 ml) and piperidine (0.1 ml) was refluxed for 4 h, then processed as in the general procedure A to give compounds **3a-j** (analytical and spectral data).

3a. Obtained as yellow crystals (ethanol), yield 81%; mp 178-179°C. IR spectrum, ν , cm^{-1} : 3433, 3213 (NH_2), 3345 (NH-imidazoline), 2178 (CN) and 1647 ($\text{C}=\text{C}-\text{NO}_2$). Found, %: C 59.63; H 4.85; N 24.85. $\text{C}_{14}\text{H}_{13}\text{N}_5\text{O}_2$. Calculated, %: C 59.36; H 4.62; N 24.72.

3b. Obtained as yellow crystals (ethanol), yield 83%; mp 205-206°C. IR spectrum, ν , cm^{-1} : 3418, 3237 (NH_2), 3341 (NH-imidazoline), 2174 (CN) and 1641 ($\text{C}=\text{C}-\text{NO}_2$). ^1H NMR spectrum, δ , ppm: 3.85 (4H, m, 2CH_2 imidazoline); 4.59 (1H, s, H-7); 6.48 (2H, br. s, NH_2); 7.12-7.38 (4H, m, aromatic); 9.38 (1H, br. s, NH imidazoline). Found, %: C 53.12; H 3.93; N 22.21. $\text{C}_{14}\text{H}_{12}\text{ClN}_5\text{O}_2$. Calculated, %: C 52.92; H 3.81; N 22.05.

3c. Obtained as yellow crystals (ethanol), yield 78%; mp 245-246 (dec.). IR spectrum, ν , cm^{-1} : 3445, 3225 (NH_2), 3333 (NH-imidazoline), 2182 (CN), 1629 ($\text{C}=\text{C}-\text{NO}_2$). Mass spectrum, m/z (*I*, %): 362 $[\text{M}-\text{C}_6\text{H}_4\text{Br}]^+$ (62.3); 206 $[\text{M}-(\text{C}_6\text{H}_4\text{Br} + \text{NO}_2)]^+$ (100); 156 $[\text{C}_6\text{H}_4\text{Br}]^+$ (7). Found, %: C 46.63; H 3.52; N 19.47. $\text{C}_{14}\text{H}_{12}\text{BrN}_5\text{O}_2$. Calculated, %: C 46.42; H 3.34; N 19.34.

3d. Obtained as yellow crystals (acetone), yield 76%; mp 235-236 dec. IR spectrum, ν , cm^{-1} : 3415, 3235 (NH_2), 3334 (NH-imidazoline), 2182 (CN), 1628 ($\text{C}=\text{C}-\text{NO}_2$). Found, %: C 50.94; H 3.52; N 25.47. $\text{C}_{14}\text{H}_{12}\text{N}_6\text{O}_4$. Calculated, %: C 51.22; H 3.69; N 25.60.

3e. Obtained as yellow crystals (acetone), yield 79%; mp 245-246 dec. IR spectrum, ν , cm^{-1} : 3385, 3325 (NH_2), 3300 (NH-pyrimidine), 2185 (CN), 1628 ($\text{C}=\text{C}-\text{NO}_2$). Mass spectrum, m/z (*I*, %): 331 (12), 285 $[\text{M}-\text{NO}_2]^+$ (34), 262 $[\text{M}-\text{C}_6\text{H}_4\text{Cl}]^+$ (100), 174 $[\text{M}-(\text{C}_6\text{H}_4\text{Cl} + \text{NO}_2)]^+$ (10). Found, %: C 54.13; H 4.12; N 21.03. $\text{C}_{15}\text{H}_{14}\text{ClN}_5\text{O}_2$. Calculated, %: C 54.29; H 4.25; N 21.11.

3f. Obtained as yellow crystals (acetone), yield 77%; mp 249-250 dec. IR spectrum, ν , cm^{-1} : 3386, 3226 (NH_2), 3300 (NH-pyrimidine), 2186 (CN), 1629 ($\text{C}=\text{C}-\text{NO}_2$). ^1H NMR spectrum, δ , ppm: 2.49-2.53 (2H, m, $3-\text{CH}_2$); 3.73-3.80 (4H, m, $2-\text{CH}_2$ and $4-\text{CH}_2$); 4.68 (1H, s, H-8); 6.48 (2H, s, NH_2); 7.11-7.49 (4H, m, aromatic); 11.55 (1H, br. s, NH). Found, %: C 48.11; H 3.92; N 18.84. $\text{C}_{15}\text{H}_{14}\text{BrN}_5\text{O}_2$. Calculated, %: C 47.89; H 3.75; N 18.62.

3g. Obtained as yellow crystals (acetone), yield 76%; mp 225-226°C dec. IR spectrum, ν , cm^{-1} : 3391, 3228 (NH_2), 3300 (NH-pyrimidine), 2184 (CN), 1625 ($\text{C}=\text{C}-\text{NO}_2$). Found, %: C 52.41; H 3.92; N 24.73. $\text{C}_{15}\text{H}_{14}\text{N}_6\text{O}_4$. Calculated, %: C 52.63; H 4.12; N 24.55.

3h. Obtained as golden yellow crystals (acetone), yield 69%; mp 244-245°C dec. IR spectrum, ν , cm^{-1} : 3418, 3223 (NH_2), 3293 (NH-diazepine), 2190 (CN), 1624 ($\text{C}=\text{C}-\text{NO}_2$). Mass spectrum, m/z (*I*, %): 345 $[\text{M}-\text{NO}_2]^+$ (6.85); 299 $[\text{M}-\text{C}_6\text{H}_4\text{Cl}]^+$ (90.8), 234 $[\text{M}-(\text{NO}_2 + \text{C}_6\text{H}_4\text{Cl})]^+$ (45.7), 188 (11.3), 55 (100). Found, %: C 55.42; H 4.37; N 20.08. $\text{C}_{16}\text{H}_{16}\text{ClN}_5\text{O}_2$. Calculated, %: 55.57; H 4.66; N 20.25.

3i. Obtained as golden yellow crystals (ethanol), yield 67 %; mp 240-242°C. IR spectrum, ν , cm^{-1} : 3400-3217 (NH_2), 3287 (NH-diazepine), 2191 (CN), 1622 ($\text{C}=\text{C}-\text{NO}_2$). ^1H NMR spectrum, δ , ppm: 2.53-2.55

(4H, m, 3-CH₂, 4-H₂); 3.73-3.82 (4H, m, 2-CH₂, 5-CH₂); 4.73 (1H, s, H-9); 6.52 (2H, s, NH₂); 7.16-7.51 (4H, m, aromatic); 11.25 (1H, br. s, NH). Found, %: C 49.48; H 4.27; N 18.11. C₁₆H₁₆BrN₅O₂. Calculated, %: C 49.25; H 4.13; N 17.95.

3j. Obtained as yellow crystals (ethanol), yield 71%; mp 235-236°C dec. IR spectrum, ν , cm⁻¹: 3385, 3216 (NH₂), 3290 (NH-diazepine), 2191 (CN), 1624 (C=C-NO₂). Found, %: C 54.19; H 4.35; N 23.30. C₁₆H₁₆N₆O₄. Calculated, %: C 53.93; H 4.53; N 23.59.

6-(Benzyl or butyl)-2,3,4,5,6,7-hexahydronitro-1H-imidazo[1,2-c]pyrimidines 6a,b. A mixture of **2a** (0.003 mol), benzylamine or *n*-butylamine (0.003 mol), and formalin (0.006 mol) in ethanol (30 ml) was refluxed for 3 h. The solid products obtained after concentration and cooling were filtered off and crystallized from ethanol to give compounds **6a,b**.

6a. Obtained as colorless crystals, yield 64%; mp 179-180°C. ¹H NMR spectrum, δ , ppm: 3.45, 3.75 (8H, m, 4 × CH₂ heterocyclic); 4.04 (2H, s, CH₂ benzylic); 7.33 (5H, m, aromatic); 8.77 (1H, s, NH-imidazoline). Found, %: C 60.17; H 6.23; N 21.79. C₁₃H₁₆N₄O₂. Calculated, %: C 59.98; H 6.20; N 21.53.

6b. Obtained as pale yellow crystals, yield 56%; mp 118-120°C. IR spectrum, ν , cm⁻¹: 3259 (NH-imidazoline), 1594 (C=C-NO₂). Found, %: C 53.24; H 8.29; N 24.93. C₁₆H₁₈N₄O₂. Calculated, %: C 53.07; H 8.02; N 24.76.

2,2'-Methylenebis[2-(nitromethylene)imidazoline] and 2,2'-Methylenebis[hexahydro-2-(nitromethylene)pyrimidine] (7a,b). A mixture of **2a** or **2b** (0.003 mol), the appropriate aliphatic secondary amine (0.003 mol), and formalin (0.003 mol) in ethanol (30 ml) was refluxed for 3 h. The solid products obtained after concentration and cooling were filtered off and crystallized from ethanol to give colorless crystals of **7a,b**.

7a. Yield 77%; mp 245-246°C (dec). IR spectrum, ν , cm⁻¹: 3357 (NH-imidazoline), 1601 (C=C-NO₂). ¹H NMR spectrum, δ , ppm: 3.81-3.87 (8H, complex pattern, 4CH₂-imidazolines), 4.10 (2H, s, CH₂ side chain), 9.23-9.29 (4H, br. s, NH-imidazolines). Found, %: C 39.82; H 5.11; N 30.92. C₉H₁₄N₆O₄. Calculated, %: C 39.99; H 5.22; N 31.10.

7b. Yield 73%; mp 255-256°C, dec. IR spectrum, ν , cm⁻¹: 3256 (NH-pyrimidine), 1613 (C=C-NO₂). Mass spectrum, m/z (*I*, %): 206 [M-2NO₂]⁺ (5), 156 [C₆H₁₀N₃O₂] (16), 110 [156-NO₂]⁺ (12) and 56 [C₃H₆N] (100). Found, %: C 44.47; H 6.23; N 28.37. C₁₁H₁₈N₆O₄. Calculated, %: C 44.29; H 6.08; N 28.18.

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